

AMENDMENTS TO THE SPECIFICATION

Please replace the paragraph beginning at page 64, line 22, with the following amended paragraph:

Benzyl alcohol (10 ml) was added to 0.2 g of a polyamide resin sample and the sample was dissolved at $205 \pm 5^\circ\text{C}$ for 5 minutes. The resulting solution was cooled in water for 15 seconds and after adding phenol phthalein as an indicator, titrated with an ethanolic potassium hydroxide solution (prepared by adding ethanol to 80 ml of 0.5N-KOH to make 1,000 ml), and the carboxyl end group concentration [CEG] was calculated according to the following formula:

$$\text{CEG (meq/kg)} = \{[(A-B) \times N \times f] / (w \times 1000)\} \times 10^6$$

A: titration value (ml)

B: blank titration value of ~~dissolution~~ solvent (ml)

N: concentration (mol/liter) of ethanolic potassium hydroxide

f: factor of ethanolic potassium hydroxide

w: weight (g) of sample

Please replace the paragraph beginning at page 89, line 4, with the following amended paragraph:

The melted raw materials, that is, ADA/CHDA mixture and MXD, were supplied by plunger pumps (15) and (16), respectively, to the tubular reaction apparatus ($L/D=780$) (21) for the amidation step each in a constant amount. At this time, as for the mass flow rates of ADA and ~~HMDA~~ CHDA, the outputs of the plunger pumps (15) and (16) were automatically controlled by the control unit (19) so that the mass flowmeters (17) and (18) (both, Model CN003D-SS-200R, Coriolis-type flowmeter, manufactured by Oval Corporation) disposed immediately after the plunger pumps (15) and (16) could indicate the values shown in Table 5, respectively.